

## A Series of Quinoline Analogues as Potent Inhibitors of *C. albicans* Prolyl tRNA Synthetase

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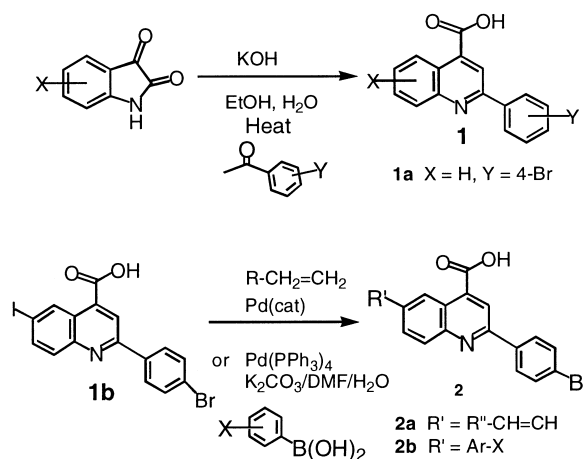
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**Abstract**—A series of quinoline inhibitors of *C. albicans* prolyl tRNA synthetase was identified. The most potent analogue, 2-(4-bromo-phenyl)-6-chloro-8-methyl-4-quinolinecarboxylic acid, showed  $IC_{50} = 5$  nM (Ca. ProRS) with high selectivity over the human enzyme. © 2001 Elsevier Science Ltd. All rights reserved.

The incidence of serious fungal infection continues to grow and there is an urgent need for novel antifungal therapies.<sup>1,2</sup> Aminoacyl-tRNA synthetases (aaRSs) are essential enzymes for biological cell growth.<sup>3–7</sup> Because there are significant structural difference between fungal and human enzymes, selective inhibition of aaRSs offers a potential pathway for discovery of novel antifungal agents. Previously, we have described novel antibacterials targeting aminoacyl-tRNA synthetases.<sup>8,9</sup> Herein, we report novel quinoline inhibitors of *C. albicans* prolyl tRNA synthetase. The quinoline lead, 2-(4-bromo-phenyl)-4-quinolinecarboxylic acid **1a**, was identified from high-throughput screening of library compounds. Compound **1a** inhibits *C. albicans* prolyl-tRNA synthetase ( $IC_{50} = 0.5$   $\mu$ M) with high selectivity over the human enzyme ( $IC_{50} > 100$   $\mu$ M). To examine the structure–activity relationships, we have prepared and evaluated a series of quinoline analogues.

Our synthetic approach for the preparation of quinoline analogues is shown in Scheme 1. Reaction of a substituted isatin with a ketone gave 4-quinoline carboxylic acid derivative **1**.<sup>10</sup> The 6-alkenyl group (**2a**) was introduced from coupling of iodoquinoline **1b** with alkene under a typical Heck condition.<sup>11</sup> The 6-aryl group (**2b**) was introduced from a palladium-catalyzed (Suzuki) coupling of arylboronic acid and iodoquinoline **1b**.<sup>12</sup>

Quinoline analogues with modification at the 8-methyl position were prepared as shown in Scheme 2. The



Scheme 1.

quinoline **1c** was esterified and converted to the bromide quinoline **3**.<sup>13</sup> Treatment of **3** with aqueous silver nitrate followed by hydrolysis gave the corresponding alcohol **4**.<sup>14</sup> Dibromination of **1c** followed by a similar treatment with aqueous silver nitrate, depending upon the conditions, yielded benzaldehyde quinoline **6** or benzoic acid quinoline **7**.

Derivatives of the 4-quinolinecarboxylic acid were prepared as shown in Scheme 3. Reduction of **1c** gave the hydroxyl derivative **8**, which was converted to the sulfamoyl derivative **9**.<sup>8</sup> Treatment of the primary amide with thionyl chloride yielded the nitrile derivative **12**,<sup>15</sup> which

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was reacted with trimethylsilyl azide in the presence of dibutyltin oxide to afford the tetrazole derivative **13**.<sup>16</sup>

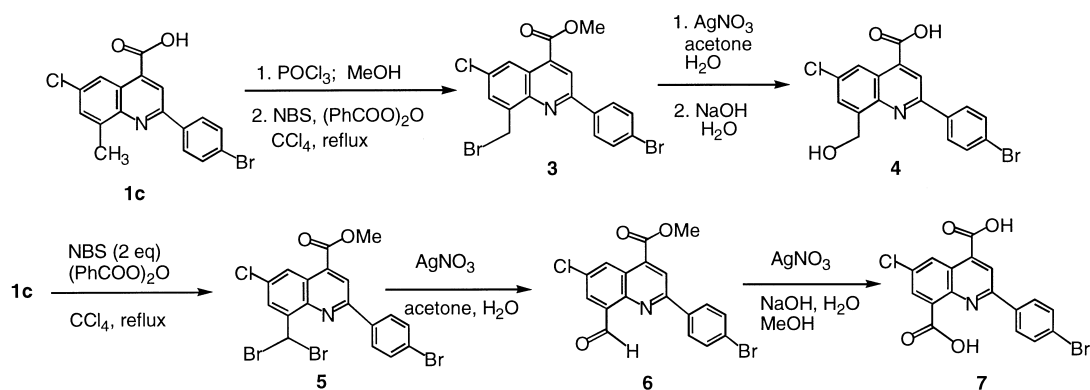
2-Aryl analogues were evaluated in assays of the aminoacylation activity of *C. albicans* prolyl-tRNA synthetase and the results are shown in Table 1. The bromophenyl compound **1a** was significantly more potent than the simple phenyl analogue **1d**. Analogue **1e** with an electron-donating -OMe group and analogue **1f** with an electron-withdrawing -CF<sub>3</sub> group showed a significant loss in potency. Replacement of the bromophenyl group by a heterocyclic group such as the 2-Cl-thiophene group (**1i**) decreased the activity. The furyl analogue **1j** and the pyridyl analogue **1k** were inactive.

Substitution in the benzene portion of quinoline can dramatically improve the activity (Table 2). 6-Halogen substituted analogues (**1b**, **1l** and **1m**) were all very active. However, incorporation of hydrophilic moieties at the 6-position (**1o** and **1p**) decreased the potency. The 6-chloro-8-methyl analogue **1c** was the most potent compound (IC<sub>50</sub> = 5 nM). The 6-chloro-8-hydroxymethylene analogue **4** (IC<sub>50</sub> = 1500 nM), however, was considerably less potent. This observation implied the importance of lipophilicity of the substitution in the 8-position for good inhibition. A comparison of analogue **1c** and analogue **1r** reveals that changing the relative position of the chloro group and the methyl group resulted in a 5-fold weaker inhibition. The 6,8-dimethyl analogue **1q** was 5000 times less potent than the 6-chloro-

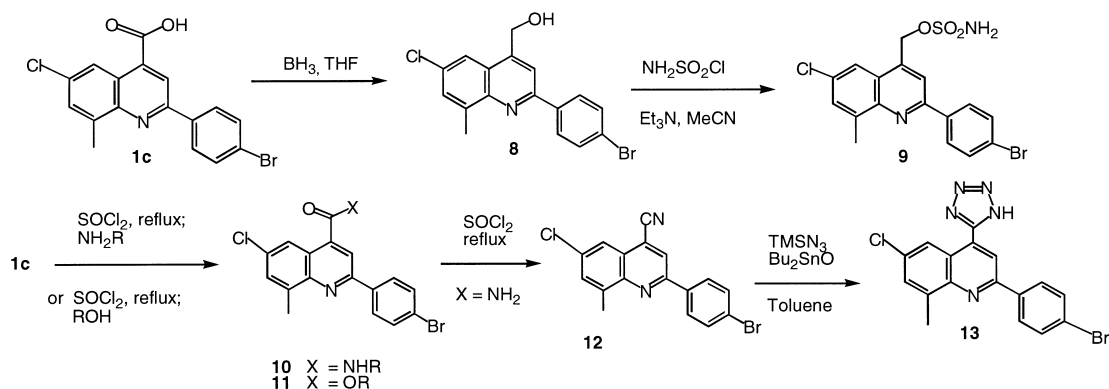
8-methyl analogue **1c**. Substitution with a halogen at the 7-position and a methyl at the 8-position (**1v**) and (**1w**) decreased the potency. All of these compounds have high selectivity over the human enzyme.

To explore other functional groups at the 4-position, we replaced the carboxylic acid with sulfamoyl amide, amide, ester, and tetrazole (Table 3). The replacement of the carboxylic acid with the methylene sulfomyl amide (**9**), the amide (**10a**) and the phenylbenzyl ester (**11a**) decreased the potency. The 3,4-dichlorobenzyl ester (**11b**), the furfuryl ester (**11c**), and the tetrazole derivative **13** had a moderate inhibition. The results suggest that a carboxylic acid functionality at this position is critical for good enzyme inhibition.

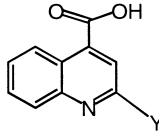
In summary, we have developed a series of potent and selective *C. albicans* prolyl-tRNA synthetase inhibitors. The enzymatic potency observed with several of these analogues is significantly superior to that of lead compound **1a** (IC<sub>50</sub> = 500 nM, *C. albicans* ProRS). SAR studies suggest that the potency is greatly influenced by the substitution pattern on the quinoline. Analogue **1c** possesses excellent activity against *C. albicans* ProRS (IC<sub>50</sub> = 5 nM) with high selectivity over the human enzyme (IC<sub>50</sub> > 100 μM). Enzymology studies indicate that compound **1c** is a non-competitive inhibitor of the *C. albicans* ProRS with respect to proline and a competitive inhibitor with respect to ATP. The K<sub>i</sub>'s were 15 and 5.1 nM versus proline and ATP, respectively. However, the












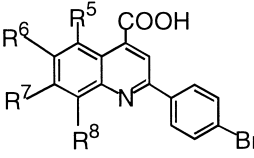
Scheme 2.



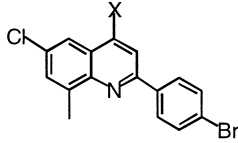
Scheme 3.

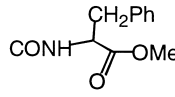
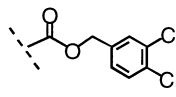
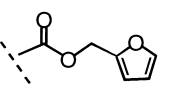

**Table 1.** Inhibition of *C. albicans* prolyl-tRNA synthetase


Y Compd									
	<b>1a</b>	<b>1d</b>	<b>1e</b>	<b>1f</b>	<b>1g</b>	<b>1h</b>	<b>1i</b>	<b>1j</b>	<b>1k</b>
IC <sub>50</sub> (μM) (Ca. ProRS)	0.5	>100	>100	>100	19	3.6	32	>100	>100
IC <sub>50</sub> (μM) (human ProRS)	>100	>100	>100	>100	>100	>100	>100	>100	>100

**Table 2.** Inhibition of *C. albicans* prolyl-tRNA synthetase


Compd	R <sup>5</sup>	R <sup>6</sup>	R <sup>7</sup>	R <sup>8</sup>	IC <sub>50</sub> (μM) (Ca. ProRS)	IC <sub>50</sub> (μM) (human ProRS)
<b>1a</b>	H	H	H	H	0.500	>100
<b>1b</b>	H	I	H	H	0.019	>100
<b>1l</b>	H	Cl	H	H	0.026	>20
<b>1m</b>	H	Br	H	H	0.025	>20
<b>1n</b>	H	CF <sub>3</sub> O	H	H	0.053	>100
<b>1o</b>	H	HOS <sub>2</sub> O	H	H	15	>100
<b>1p</b>	H	NH <sub>2</sub>	H	H	3.3	95
<b>2a</b>	H	HO <sub>2</sub> CCH=CH <sub>2</sub>	H	H	1.3	80
<b>2b</b>	H	4'-OH-PH	H	H	0.350	35
<b>1c</b>	H	Cl	H	Me	<b>0.005</b>	<b>&gt;20</b>
<b>1q</b>	H	Me	H	Me	25	>100
<b>1r</b>	H	Me	H	Cl	0.025	>20
<b>4</b>	H	Cl	H	CH <sub>2</sub> OH	1.5	>100
<b>1s</b>	Me	H	Cl	H	1.5	>100
<b>1t</b>	Cl	H	Cl	H	0.140	>100
<b>1u</b>	H	Br	H	CF <sub>3</sub>	0.220	56
<b>1v</b>	H	H	Br	Me	0.150	70
<b>1w</b>	H	H	F	Me	0.120	60

**Table 3.** Inhibition of *C. albicans* prolyl-tRNA synthetase


X Compd	CO <sub>2</sub> H	CH <sub>2</sub> OSO <sub>2</sub> NH <sub>2</sub>		CO <sub>2</sub> CH <sub>2</sub> PhPh			
	<b>1c</b>	<b>9</b>	<b>10a</b>	<b>11a</b>	<b>11b</b>	<b>11c</b>	<b>13</b>
IC <sub>50</sub> (Ca. ProRS, μM)	0.005	2.4	5.7	8.8	0.02	0.07	0.03
IC <sub>50</sub> (human ProRS, μM)	>100	—	>100	>100	>100	>100	>100

most active analogue **1c** has no whole cell activity (MIC >100 μg/mL, *C. albicans*). Analogue **1v** has cellular activity (MIC = 25 μg/mL, *C. albicans*). Further studies indicate that the activity is due to a lytic mechanism.

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